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Diastereoselective Synthesis of CF3‑Substituted, Epoxide-Fused Heterocycles with β -(Trifluoromethyl)vinylsulfonium Salts

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 CF_3 -substituted vinyl diphenylsulfonium triflate is an effective annulation reagent for the formation of α -CF₃ substituted, epoxide-fused heterocycles (pyrrolidines, piperidines, and tetrahydrofurans). This simple method affords a variety of valuable heterocyclic building blocks in a highly diastereoselective manner (dr >20:1).

Fluorinated and especially CF_3 -substituted compounds are of considerable contemporary interest, $\frac{1}{x}$ due to the development of biologically active compounds containing this functionality.2 Combined into heterocyclic frameworks, this often leads to the creation of superior pharmacophores.³ Despite this high interest, efficient methods, starting from simple materials, for the introduction of the trifluoromethyl group into saturated heterocycles are still scarce.4,5 We recently reported the synthesis of epoxide- and aziridinefused five-, six-, and seven-membered heterocycles from unsubstituted vinylsulfonium triflates $6,7$ and explored alternative modes of reactivity with these reagents for the construction of other monocyclic four- to seven-membered heterocycles such as morpholines.^{8,9} We envisioned that we could combine our advances on the Michael-type-addition/ annulation sequence for fused ring systems with CF_3 substituted vinylsulfonium salts reported by others¹⁰ to produce useful heterocyclic building blocks (Scheme 1).

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^{(1) (}a) Yale, H. L. J. Med. Pharmaceut. Chem. 1959, 1, 121. (b) Dolbier, W. R. J. Fluorine Chem. 2005, 126, 157. (c) Muller, K.; Faeh, C.; Diederich, F. Science 2007, 317, 1881. (d) O'Hagan, D. Chem. Soc. Rev. 2008, 37, 308.

^{(2) (}a) Jeschke, P. ChemBioChem 2004, 5, 5701. (b) Begue, J. P.; Bonnet-Delpon, D. J. Fluorine Chem. 2006, 127, 992. (c) Purser, S.; Moore, P. R.; Swallow, S.; Gouverneur, V. Chem. Soc. Rev. 2008, 37, 320. (d) Hagmann, W. K. J. Med. Chem. 2008, 51, 4359.

⁽³⁾ For a selection of recent examples, see: (a) Chaume, G.; Van Severen, M. C.; Marinkovic, S.; Brigaud, T. Org. Lett. 2006, 8, 6123. (b) Bezdudny, A. V.; Alekseenko, A. N.; Mykhailiuk, P. K.; Manoilenko, O. V.; Shishkin, O. V.; Pustovit, Y. M. Eur. J. Org. Chem. 2011, 1782. (c) Kenis, S.; D'hooghe, M.; Verniest, G.; Nguyen, V. D.; Tuyet, A. D. T.; Nguyen, T. V.; De Kimpe, N. Org. Biomol. Chem. 2011, 9, 7217. (d) Maeda, R.; Ishibashi, R.; Kamaishi, R.; Hirotaki, K.; Furuno, H.; Hanamoto, T. Org. Lett. 2011, 13, 6240. (e) Kenis, S.; D'hooghe, M.; Verniest, G.; Tuyet, A. D. T.; The, C. P.; Nguyen, T. V.; De Kimpe, N. J. Org. Chem. 2012, 77, 5982.

^{(4) (}a) Shimizu, M.; Hiyama, T. Angew. Chem., Int. Ed. 2005, 44, 214. (b) Fustero, S.; Sanz-Cervera, J. F.; Acena, J. L.; Sanchez-Rosello, M. Synlett 2009, 525. (c) Furuya, T.; Kamlet, A. S.; Ritter, T. Nature 2011, 473, 470. (d) Nie, J.; Guo, H.-C.; Cahard, D.;Ma, J.-A. Chem. Rev. 2011, 111, 455. (e) Bariau, A.; Jatoi, W. B.; Calinaud, P.; Troin, Y.; Canet, J.-L. Eur. J. Org. Chem. 2006, 3421. (f) Lin, P.; Jiang, J. Tetrahedron 2000, 56, 3635.

⁽⁵⁾ For recent examples on the formation of sp^3 C-CF₃ bonds leading to building blocks for similar heterocycles, see: (a) Kieltsch, I.; Eisenberger, P.; Togni, A. Angew. Chem., Int. Ed. 2007, 46, 754. (b) Pham, P. V.; Nagib, D. A.; MacMillan, D. W. C. Angew. Chem., Int. Ed. 2011, 50, 6119. (c) Yasu, Y.; Koike, T.; Akita, M. Angew. Chem., Int. Ed. 2012, 51, 9567 and references therein.

Screening of reaction conditions using aminoketone 2a and β-trifluoromethylvinyl sulfonium salt 1 led to an optimized method that produced 3a in 81% yield and in >20:1 dr (please see Supporting Information for optimization details). The scope of the reaction (Table 1) was shown to extend to the synthesis of other N-tosyl pyrrolidines $3a-d$, giving good yields and excellent diastereoselectivity. Sulfonamide 3e bearing the easier-to-cleave p -Ns¹¹ also worked well. Furthermore, the synthesis of fused

 (7) For our work, see: (a) Unthank, M. G.; Hussain, N.; Aggarwal, V. K. Angew. Chem., Int. Ed. 2006, 45, 7066. (b) Kokotos, C. G.; McGarrigle, E. M.; Aggarwal, V. K. Synlett 2008, 219. (c) Unthank, M. G.; Tavassoli, B.; Aggarwal, V. K. Org. Lett. 2008, 10, 1501. (d) Yar, M.; Unthank, M. G.; McGarrigle, E. M.; Aggarwal, V. K. Chem. - Asian. J. 2011, 6, 372. (e) Fritz, S. P.; Ali, Z.; Unthank, M. G.; McGarrigle, E. M.; Aggarwal, V. K. Helv. Chim. Acta 2012, in press, doi:10.1002/hlca.201200455.

(8) For further vinylsulfonium mediated annulation reactions, see: (a) Matsuo, J.; Yamanaka, H.; Kawana, A.; Mukaiyama, T. Chem. Lett. 2003, 32, 392. (b) Yamanaka, H.; Matsuo, J.; Kawana, A.; Mukaiyama, T. Chem. Lett. 2003, 32, 626. (c) Yamanaka, H.; Mukaiyama, T. Chem. Lett. 2003, 32, 1192. (d) Yamanaka, H.; Matsuo, J.; Kawana, A.; Mukaiyama, T. ARKIVOC 2004, 42. (e) Yar, M.; McGarrigle, E. M.; Aggarwal, V. K. Angew. Chem., Int. Ed. 2008, 47, 3784. (f) Hansch, M.; Illa, O.; McGarrigle, E. M.; Aggarwal, V. K. Chem.— Asian. J. 2008, 3, 1657. (g) Yar, M.; McGarrigle, E. M.; Aggarwal, V. K. Org. Lett. 2009, 11, 257. (h) Xie, C. S.; Han, D. Y.; Liu, J. H.; Xie, T. Synlett 2009, 3155. (i) Bornholdt, J.; Felding, J.; Kristensen, J. L. J. Org. Chem. 2010, 75, 7454. (j) Catalan-Munoz, S.; Muller, C. A.; Ley, S. V. Eur. J. Org. Chem. 2010, 183. (k) Xie, C. S.; Han, D. Y.; Hu, Y.; Liu, J. H.; Xie, T. A. Tetrahedron Lett. 2010, 51, 5238. (l) Chen, J. R.; An, J.; Chang, N. J.; Song, L. D.; Jin, Y. Q.; Ma, Y.; Xiao, W. J. Chem. Commun. 2011, 47, 1869. (m) Fritz, S. P.; Mumtaz, A.; Yar, M.; McGarrigle, E. M.; Aggarwal, V. K. Eur. J. Org. Chem. 2011, 3156. (n) McGarrigle, E. M.; Fritz, S. P.; Favereau, L.; Yar, M.; Aggarwal, V. K. Org. Lett. 2011, 13, 3060. (o) Yar, M.; Fritz, S. P.; Gates, P. J.; McGarrigle, E. M.; Aggarwal, V. K. Eur. J. Org. Chem. 2012, 160. (p) Fritz, S. P.; Moya, J. F.; Unthank, M. G.; McGarrigle, E. M.; Aggarwal, V. K. Synthesis 2012, 44, 1584. (q) Mao, Z.; Qu, H.; Zhao, Y.; Lin, X. Chem. Commun. 2012, 48, 9927.

(9) For reviews, see: (a) Fritz, S. P. Synlett 2012, 23, 480. (b) Yar, M.; McGarrigle, E. M.; Aggarwal, V. K. Sulfonium, Ethenyldiphenyl-, 1,1,1- Trifluoromethanesulfonate. In Encyclopedia of Reagents for Organic Synthesis, eEROS; Paquette, L. A., Crich, D., Fuchs, P. L., Molander, G. A., Eds.; John Wiley & Sons Ltd.: 2012; doi: 10.1002/047084289X.rn01409.

Scheme 1. Context of Presented Work Table 1. Substrate Scope of the Annulation Reaction

^a Isolated yield after purification. $\frac{b}{b}$ From polymeric aldehyde 2b. $\frac{c}{b}$ 36 h reaction time. ^dA weaker (pyridine) or stronger (NaH) base did not change the outcome of the reaction, but led to much lower conversion. ^e Combined yield of $5a$ and $5b$, $5a:5b = 2.9:1$.

piperidines 3f,g was possible, as long as enolizable protons were not present, otherwise competing elimination occurred.¹²

⁽⁶⁾ For initial studies, see: (a) Wang, Z.; Jimenez, L. S. J. Am. Chem. Soc. 1994, 116, 4977. (b) Wang, Z.; Jimenez, L. S. Tetrahedron Lett. 1996, 37, 6049. (c) Dong, W. T.; Jimenez, L. S. J. Org. Chem. 1999, 64, 2520. (d) Wang, Y. F.; Zhang, W. H.; Colandrea, V. J.; Jimenez, L. S. Tetrahedron 1999, 55, 10659. (e) Kim, K. H.; Jimenez, L. S. Tetrahedron: Asymmetry 2001, 12, 999.

Epoxide-fused tetrahydrofuran 3h was synthesized in good yield and excellent diastereoselectivity, but the attempted synthesis of six-membered oxygen heterocycles was dominated by competing elimination, forming enol ether 4.¹² Thiol $2i$, with increased acidity adjacent to the CF_3 group, also resulted in elimination giving rise to products 5a/5b. Evidently, in cases of slow cyclization or increased acidity of the CHCF₃, competing elimination dominates.¹³ The *cis* relative stereochemistry of the product from 2a was confirmed from a crystal structure of 3a (Figure 1).

Figure 1. X-ray crystal structure of 3a, showing *cis* configuration of epoxide to CF_3 (thermal ellipsoids are drawn at the 50% probability level).

Scheme 2. Proposed Reaction Pathway for the Diastereoselective Formation of 3a

(10) (a) Maeda, R.; Ooyama, K.; Anno, R.; Shiosaki, M.; Azema, T.; Hanamoto, T. Org. Lett. 2010, 12, 2548. (b) Lin, H.; Shen, Q. L.; Lu, L. J. Org. Chem. 2011, 76, 7359. (c) Kasai, N.; Maeda, R.; Furuno, H.; Hanamoto, T. Synthesis 2012, 44, 3489. Sulfonium salt 1 can be synthesized in two steps from 2-bromo-3,3,3-trifluoroprop-1-ene (ref 10c).

Based on our experience with unsubstituted vinylsulfonium salts, $⁷$ we propose that, after initial addition of the</sup> nucleophile 2a to vinyl sulfonium salt 1 to form sulfur ylide **6**, two possible pathways (**A** and **B**) leading to *cis* and *trans* products should be considered (Scheme 2). Betaine 7 would suffer less from steric strain than 8 and has more favorable dipole interactions than 8. Pathway A with a transition state leading from conformer 6A to betaine 7 would be expected to be lower in energy than pathway \bf{B} (with a transition state leading from conformer 6B to betaine 8) as the transition states will experience similar steric and dipole interactions as the betaines. This accounts for the preferred formation of the cis-epoxide 3a.

The strength of the diastereocontrol provided by the $CF₃$ group (reagent control) was probed using the chiral substrates derived from alanine, 9a,b. Out of four possible products, only two were obtained, 10 and 11, in a 1:1 ratio (Scheme 3a). Unsurprisingly, the stereogenic center in 9 does not influence which face of the vinyl sulfonium salt is attacked. The chiral substrate 9a shows an inherent preference for formation of the epoxide *cis* to the methyl group (Scheme 3b).^{7c} Thus, product 10, with the epoxide *cis* to both the Me and CF_3 groups, is "doubly matched" and expected to be easily formed. In contrast, compound 11 has the epoxide cis to the CF_3 group but *trans* to the Me group and so is mismatched. Despite the inherent bias of a 4.6:1 ratio against its formation imposed by the Me group, it was still formed with complete exclusion of the other mismatched isomer 12 [with the epoxide *cis* to the Me group (matched) but *trans* to the CF_3 group (mismatched)], showing that the CF_3 group appears to induce a selectivity of $> 92:1$ ($> 20:1 \times 4.6:1$).¹⁴

Finally, to give an example of how this method enables rapid access to functionalized building blocks with diverse options for elaboration, a regioselective opening¹⁵ of

^{(11) (}a) Kocienski, P. J. Protecting Groups; Thieme: New York, 1994. (b) Greene, T. W.; Wuts, P. G. M. Protective Groups in Organic Synthesis, 3rd ed.; Wiley-Interscience: New York, 1999.

Scheme 4. Ring Opening of 3b with $NaN₃$

epoxide 3b with NaN_3 was carried out to give azido alcohol 14 in 61% yield (Scheme 4).

In conclusion, we have demonstrated an easy and efficient synthesis of CF₃-substituted heterocyclic building blocks in good yields and very high diastereoselectivities.

(12) Substrates with an enolizable proton are prone to a competing elimination reaction. We note that with unsubstituted vinyldiphenylsulfonium salts the conjugate addition/annulation sequence was successful but it was a problem with hindered chiral vinylsulfonium salts; see ref 7a.

(13) If cyclization towards the betaine is slow, competing elimination dominates. Related competing eliminations were also observed by Hanamoto et al. (ref 10c).

Through probing matched and mismatched stereoisomers, it was found that the diastereoselectivities are >20:1 and could be higher than $> 90:1$. The methodology has been extended to an array of different classes of CF_3 substituted, epoxide-fused heterocycles (N-, O-, five and six rings), which are useful intermediates in synthesis.

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Supporting Information Available. Experimental procedures, compound characterization, spectra, and X-ray crystallographic data. This material is available free of charge via the Internet at http://pubs.acs.org.

⁽¹⁴⁾ If, for example, the CF_3 group exerted a selectivity of 50:1, then in the mismatched case we would expect to see an 11:1 ratio of 11:12, which would be observable by ${}^{1}H$ NMR. The absence of 12 shows that the CF₃ group exerts very high selectivity and probably $> 90:1$.

^{(15) (}a) Herdeis, C.; Aschenbrenner, A.; Kirfel, A.; Schwabenländer, F. Tetrahedron: Asymmetry 1997, 8, 2421. (b) Curtis, K. L.; Evinson, E. L.; Handa, S.; Singh, K. Org. Biomol. Chem. 2007, 5, 3544. (c) Rives, A.; Génisson, Y.; Faugeroux, V.; Zedde, C.; Lepetit, C.; Chauvin, R.; Saffon, N.; Andrieu-Abadie, N.; Colié, S.; Levade, T.; Baltas, M. *Eur. J. Org. Chem.* 2009, 2474.

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